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**Exercise 1** *Specific heat of metals*

In metals, specific heat has contribution both from electrons and phonons.

- At temperature  $T$  much lower than the Debye temperature, express the ratio between phononic and electronic specific heat ( $C_{ph}/C_{el}$ ) of a three-dimensional metal with Fermi temperature  $T_F$  and Debye temperature  $\Theta_D$ .
- Calculate  $C_{ph}/C_{el}$  at  $T = 1$  K and 10 K using typical values for metals:  $T_F = 50000$  K and  $\Theta_D = 300$  K. Discuss qualitatively the temperature dependence of  $C_{ph}/C_{el}$ .

**Exercise 2** *Electronic specific heat in two dimensions*

Layered crystal structures often have electronic structures that can approximately be considered two-dimensional. The high-temperature superconductor  $Tl_2Ba_2CuO_{6+\delta}$  is one such example.

- The electronic heat capacity is given by  $C_{el} = \gamma T$ . Show that in two dimensions the Sommerfeld parameter  $\gamma$  can be written as  $\gamma = \frac{A\pi k_B^2}{3\hbar^2} m$  where  $m$  is the electronic mass and  $A$  is the total area. What is the unit of  $C_{el}$ ?
- The crystal structure of  $Tl_2Ba_2CuO_{6+\delta}$  consists of stacked layers of  $CuO_2$ . Within a layer, the  $CuO_2$  forms a square lattice with a Cu-O lattice distance of  $a = 3.8 \text{ \AA}$ . The electronic **specific** heat capacity is measured in units  $J \text{ mol}^{-1} \text{ K}^{-1}$ . Show that  $\gamma = \frac{N_A a^2 \pi k_B^2}{3\hbar^2} m$  where  $N_A$  is the Avogadro number.
- $Tl_2Ba_2CuO_{6+\delta}$  is observed to have the Sommerfeld parameter  $\gamma = 6 \text{ mJ mol}^{-1} \text{ K}^{-2}$ . Using the result of (b), what is the electronic mass  $m$  for  $Tl_2Ba_2CuO_{6+\delta}$ ? How does it compare to the free electron mass?

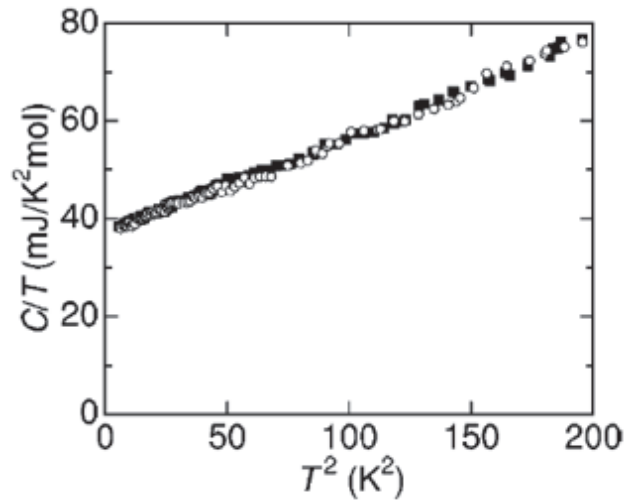


Figure 1: The total specific heat divided by temperature of  $\text{Sr}_2\text{RuO}_4$  between  $T_c$  and 14 K in zero field (filled squares) and a magnetic field of 14 T (open circles) applied parallel to the c-axis.

**Exercise 3** *Extraction of electronic and phononic specific heat*

In figure 1 the data from a specific heat experiment on  $\text{Sr}_2\text{RuO}_4$  is shown (adapted from Mackenzie et al. JPSJ **67**, 385 (1998)).

- Extract the electronic Sommerfeld parameter  $\gamma$  and the phonon coefficient  $A$  in  $C_{\text{ph}} = AT^3$ .
- The crystal structure of  $\text{Sr}_2\text{RuO}_4$  is similar to the one of  $\text{Tl}_2\text{Ba}_2\text{CuO}_{6+\delta}$ . Which of the systems would have the larger electronic mass  $m$ ?
- What is the Debye temperature for  $\text{Sr}_2\text{RuO}_4$ ?

**Exercise 4** *Specific heat of copper*

Copper has a density of  $\rho = 8.94 \text{ g cm}^{-3}$  and a molar mass of  $m_{\text{mol}} = 63.55 \text{ g mol}^{-1}$ . Use the measured values for the specific heat of copper given below to:

- determine the electron mass. Compare this value to the literature value for a free electron.
- determine the Debye temperature of copper.

$T[\text{K}] =$

0.25, 0.50, 0.75, 1.00, 1.25, 1.50, 1.75, 2.00, 2.25, 2.50, 2.75, 3.00, 3.25, 3.50, 3.75, 4.00, 4.50

$C_V[\text{mJ mol}^{-1} \text{ K}^{-1}] =$

0.17, 0.35, 0.54, 0.74, 0.96, 1.21, 1.47, 1.78, 2.11, 2.50, 2.91, 3.35, 3.91, 4.46, 5.15, 5.87, 7.49