

PHY 117 HS2023

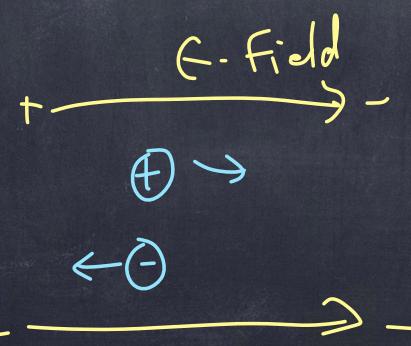
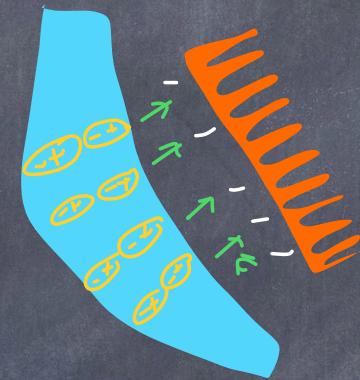
Week 9, Lecture 1

Nov. 14th, 2023

Prof. Ben Kilminster

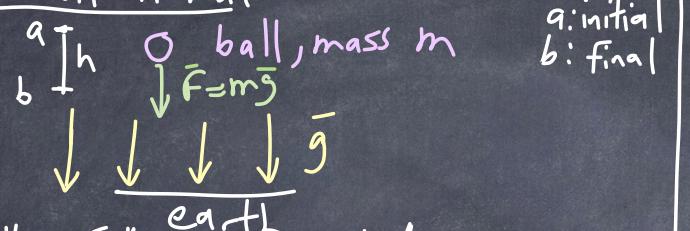


why?



Potential Energy

Gravitational



As ball falls, potential energy decreases
 $U_a > U_b$
 $mgh_a > mgh_b$

$$\Delta U = U_b - U_a = -mgh$$

The work done by gravity is $-\Delta U$

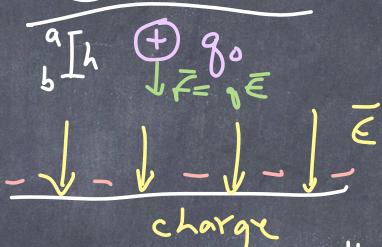
$$W_{a \rightarrow b} = -\Delta U = mgh$$

Remember

$$W_{a \rightarrow b} = \int_a^b \bar{F} \cdot d\bar{l} = +mgh$$

(+) ↓ ↓

Electrical



As (+) charge falls, potential energy decreases.
 $U_a > U_b$

$$q_0 \epsilon_a > q_0 \epsilon_b$$

$$\Delta U = U_b - U_a = -q_0 \epsilon_b h$$

The work done by \bar{E} -field is $-\Delta U$

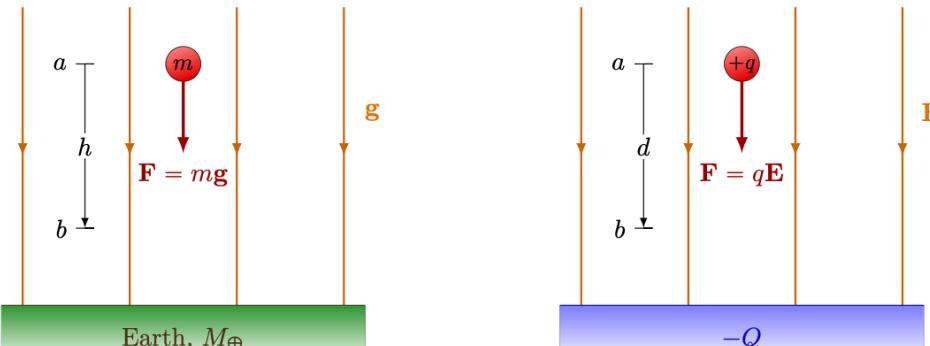
$$W_{a \rightarrow b} = -\Delta U = +q_0 \epsilon_b h$$

$$W_{a \rightarrow b} = \int_a^b \bar{F} \cdot d\bar{l} = [\bar{F} \cdot \bar{l}]_a^b = Fb - Fa$$

$$F = q_0 E$$

$$W_{a \rightarrow b} = q_0 \epsilon_b h$$

3.1 Electric potential energy



(a) Gravitational: $\Delta U = -mgh$.

(b) Electric: $\Delta U = -qEd$.

Figure 3.1: Comparison of potential energy difference $\Delta U = U_b - U_a$ in a force field.

when the movement is in the same direction as the force, there is a decrease in U .

We often use the electric potential, V , or the electric potential difference, ΔV .

$$\Delta V = V_b - V_a = \frac{U_b - U_a}{q_0} = \frac{\Delta U}{q_0} \left(= -\frac{q_0 \epsilon h}{q_0} \right)$$

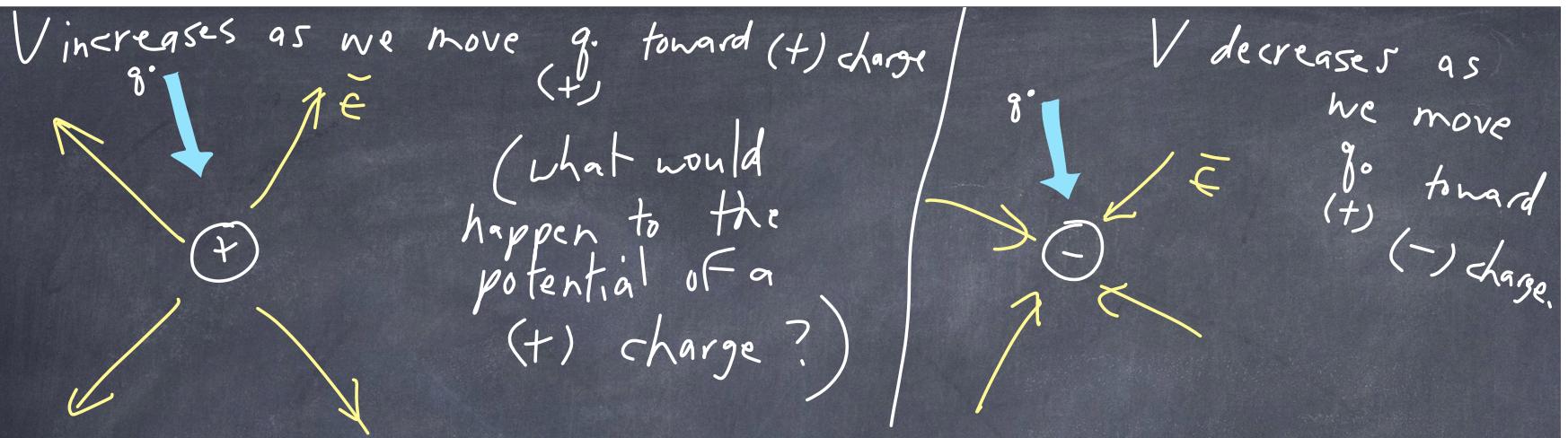
For instance
from previous
page

The electric potential is independent of the test charge,

$$dV = -\bar{E} \cdot d\bar{l}$$

$$\Delta V = - \int_a^b \bar{E} \cdot d\bar{l}$$

The (-) sign means that ΔV is (-) when movement is in the same direction as the \bar{E} -field.



The units for electric potential are Volts

$$| V = | \text{Volt} = | \left[\frac{\text{J}}{\text{C}} \right] \left[\frac{\text{energy}}{\text{charge}} \right]$$

$$\Delta V = \frac{\Delta U}{q} = - \int \vec{F} \cdot d\vec{l} = \left[\frac{\text{units}}{\text{C}} \right] = V$$

$$\Delta U = q \Delta V$$

$$[\text{J}] = [\text{C}] \left[\frac{\text{J}}{\text{C}} \right]$$

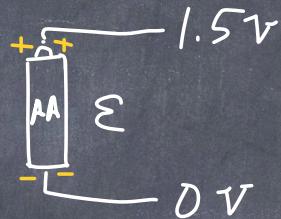
$$\bar{E} : \left[\frac{\text{N}}{\text{C}} \right] = \left[\frac{\text{V}}{\text{m}} \right]$$

$$| V = | \left[\frac{\text{J}}{\text{C}} \right] = \left[\frac{\text{N} \cdot \text{m}}{\text{C}} \right]$$

we can make a potential difference with a chemical battery

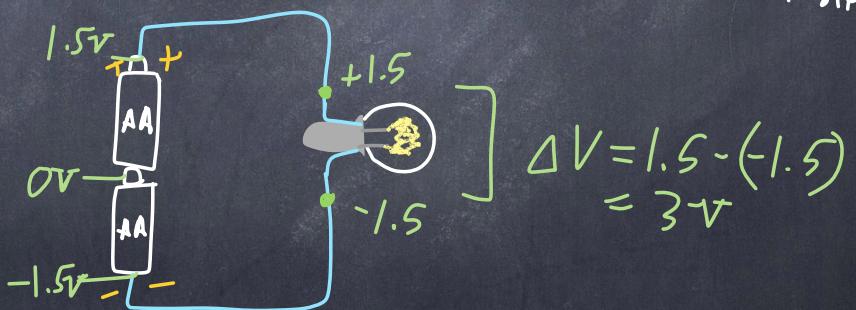
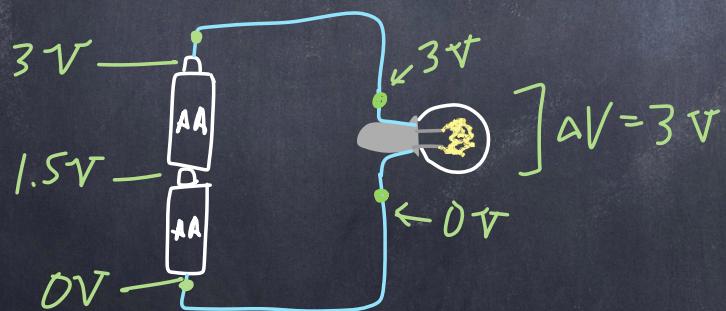
ϵ or emf
is a source
of potential
difference

we can define
0V to be
anywhere, and
we often put it
at the negative
electrode

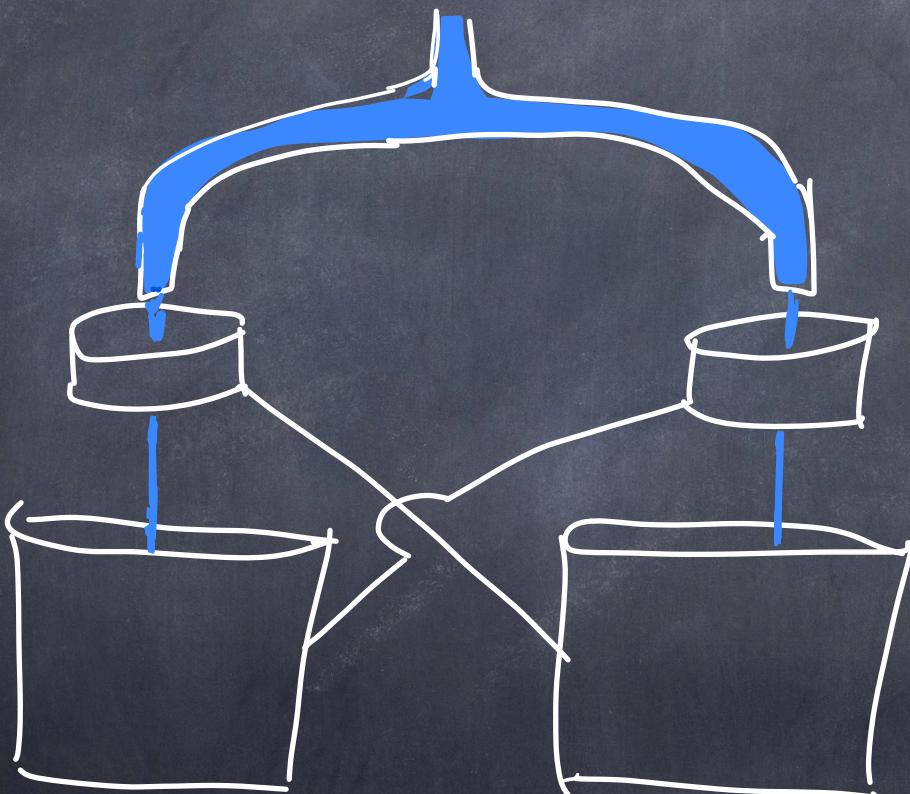


we say
 $\Delta V = 1.5V$

Electric potential is the same everywhere on a conductor. The difference in voltage (across a battery) is what defines the movement of charge + work that can be done.

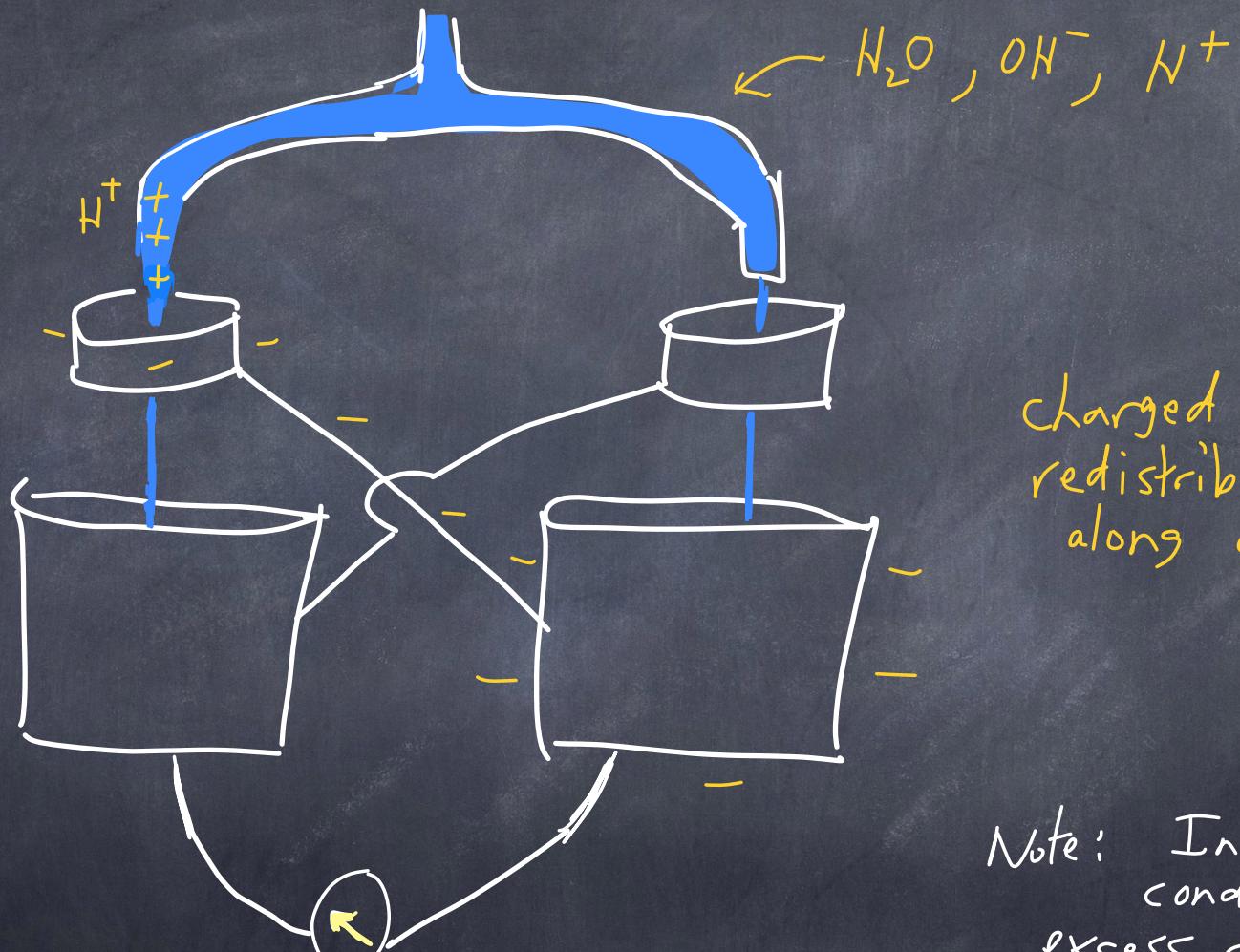


we can charge one conductor with respect to another, to create a potential difference.



Randomly, a tiny piece of dust that is charged will come along

Kelvin generator (Kelvin water dropper)

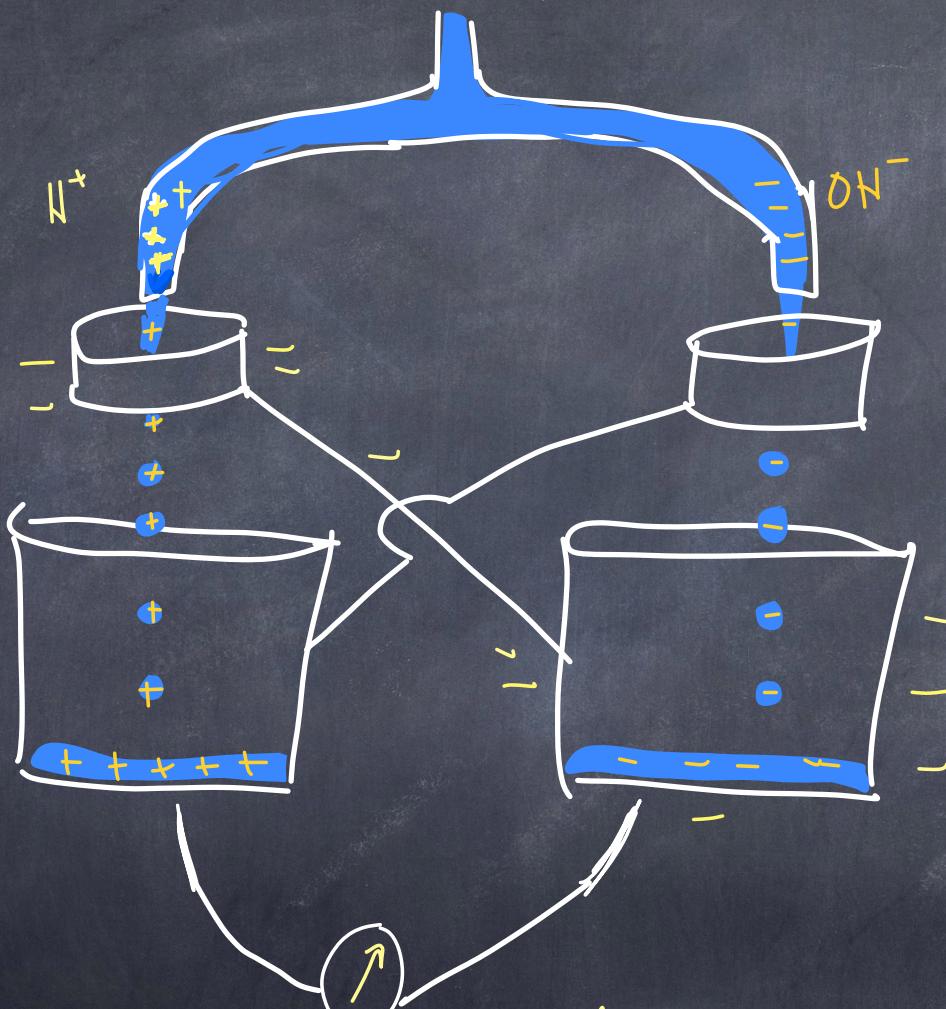


charged dust
redistributes
along conductor.

measure the
electric potential difference
between the two pairs

Note: In a
conductor,
excess charge
accumulates
on surface.

Kelvin generator (Kelvin water dropper)

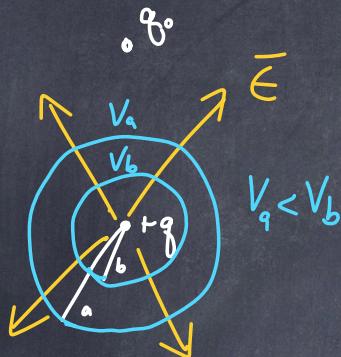


Feedback of
induced charge

Voltage difference
gets larger
& larger

Measure
The voltage difference
(relates to charge)

Potential energy due to point charge.



$$\bar{E} = \frac{kq}{r^2} \hat{r}$$

$$d\bar{E} = dr \hat{r}$$

$$dV = -\bar{E} \cdot d\bar{E}$$

$$dV = -\frac{kq}{r^2} \hat{r} \cdot dr \hat{r} = -\frac{kq}{r^2} dr$$

$$V = \int dV = \int -\frac{kq}{r^2} dr = \frac{kq}{r} + V_0 \quad \begin{matrix} \text{constant} \\ \text{of integral} \end{matrix}$$

The convention is that the potential is 0 when we are infinitely far away.

So $V = \frac{kq}{r}$ assuming $V=0$ at $r=\infty$

for a point charge

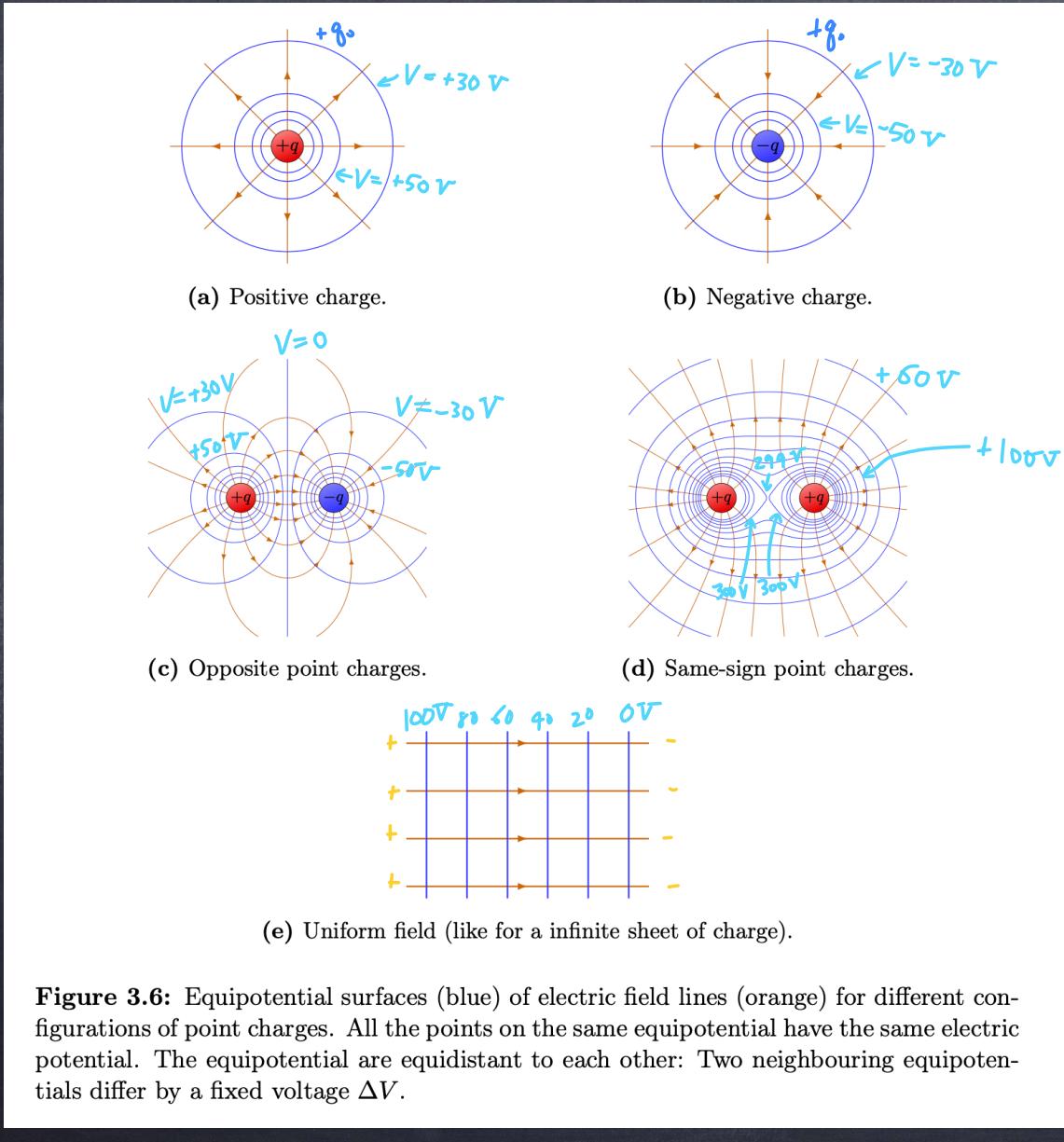
If q_0 is released from b , it will move outward from b to a

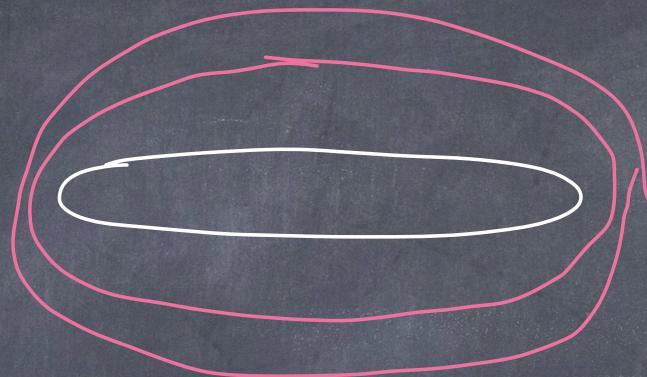
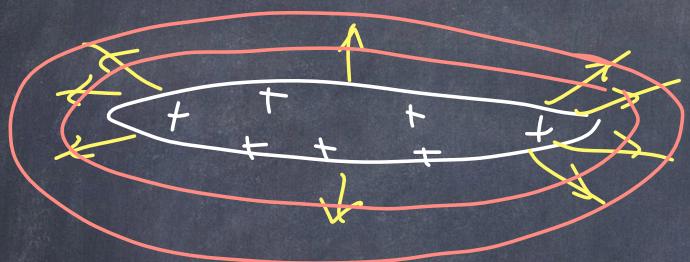
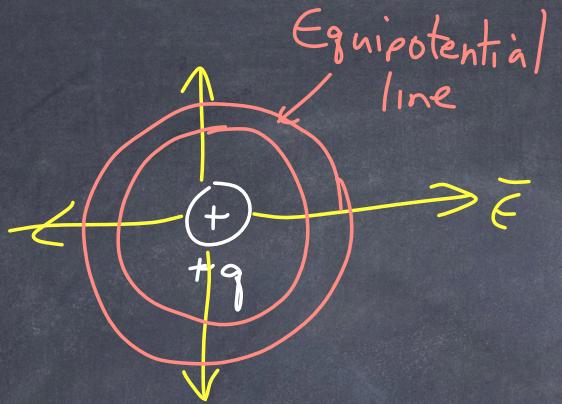
$$V_{\text{final}} - V_{\text{initial}} = V_a - V_b = \Delta V$$

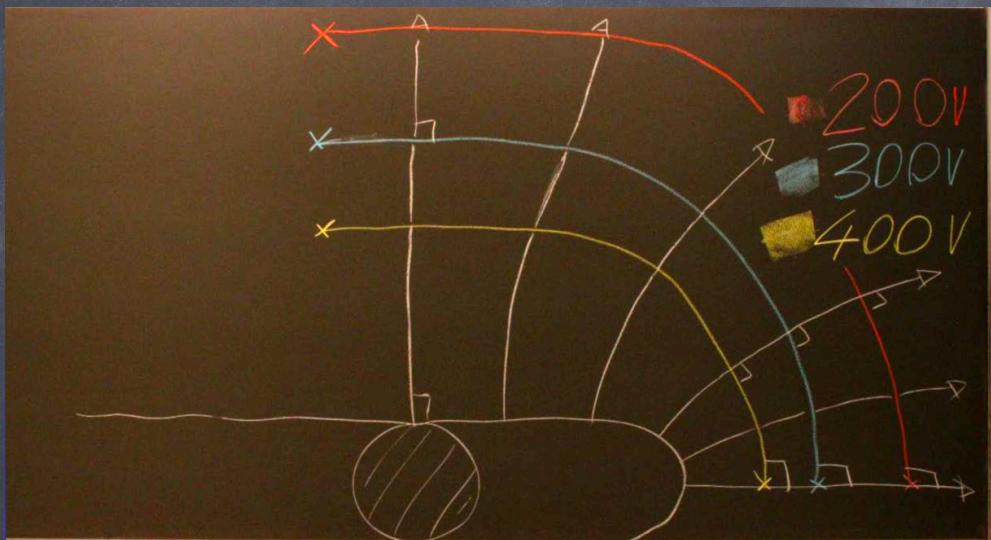
$$V_a - V_b = \frac{kq}{a} - \frac{kq}{b} = (-)$$

Decrease in potential

Equipotential lines: lines of equal potential







How do we get \bar{E} from V ?

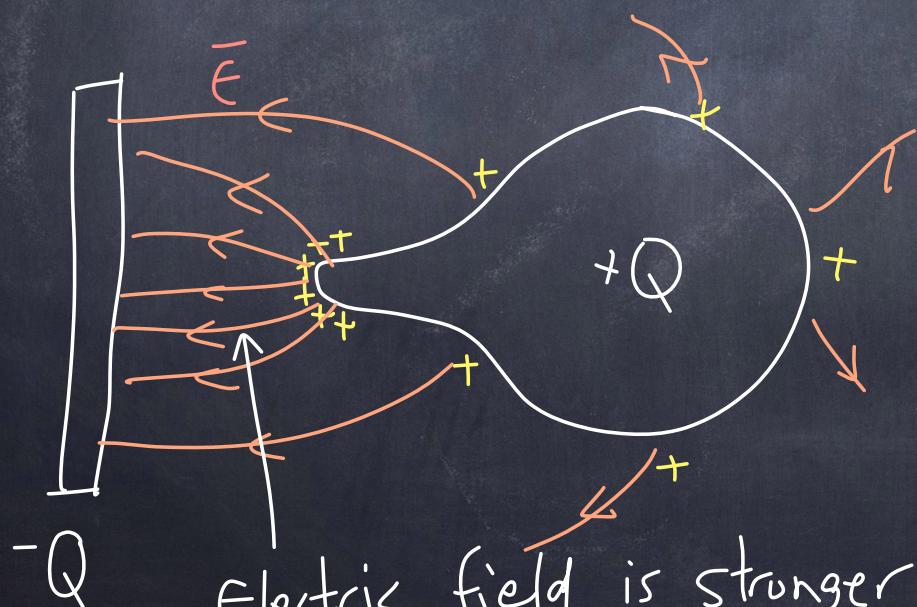
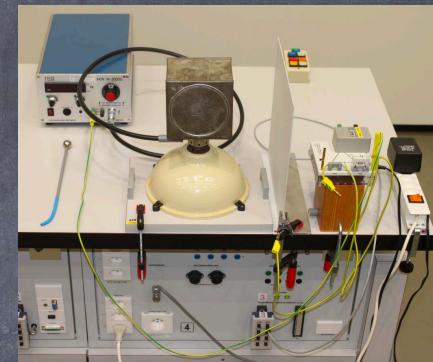
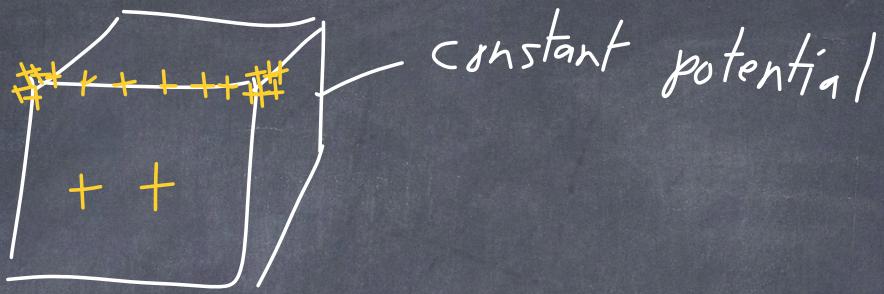
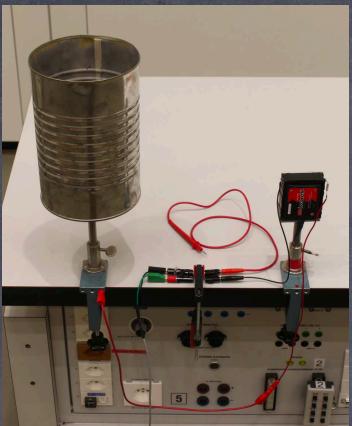
$$V = \int dV = -\int \bar{E} \cdot d\bar{l}$$



$$\bar{E} = -\frac{dV}{dl} \quad (\text{If } \bar{E} \text{ changes with } l)$$

For instance $V = \frac{kq}{r}$ (we have r changing
so we use $\frac{d}{dr}$)

$$\bar{E} = -\frac{d}{dr} \left(\frac{kq}{r} \right) \hat{r} = \frac{kq}{r^2} \hat{r}$$

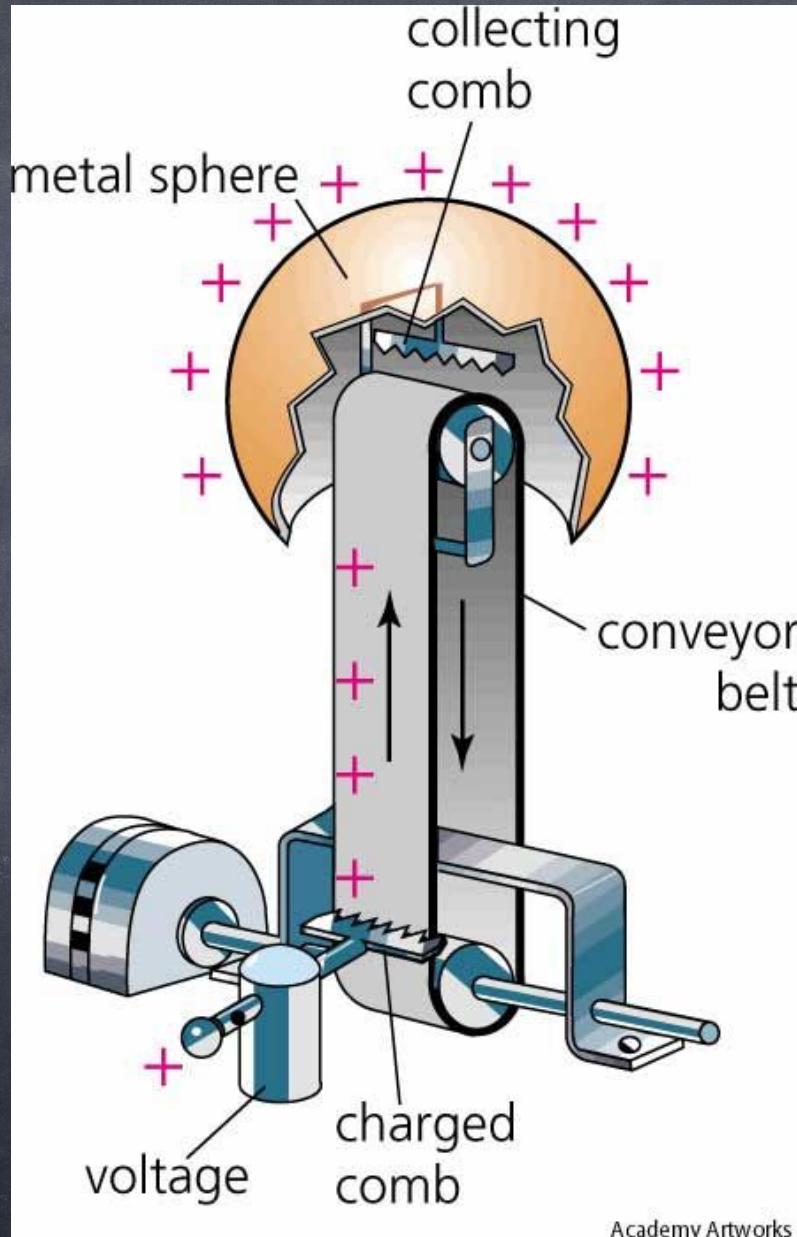


Electric field is stronger at sharper parts of object.

Van de Graaff Voltage generator

Here we add
charge to a
conductor

+
increase its
potential



Academy Artworks

Is there a limit to the potential V on a conductor from adding charge Q ?
 Yes. At high electric fields, the air becomes ionized.

$$\epsilon_{\max} \approx 3 \times 10^6 \frac{V}{m} (\text{air})$$

This is the dielectric breakdown of air, Air starts conducting electricity above this value. Lightning is electric

What is the maximum $Q + V$ of discharge. conductor with radius R in air?

$$\epsilon = \frac{\sigma}{\epsilon_0} \Rightarrow \sigma_{\max} = \epsilon_{\max} \epsilon_0$$

$$\text{For a sphere, } \sigma = \frac{Q}{4\pi R^2}$$

ϵ -Field at the surface of a conductor where

$$\sigma = \frac{Q}{\text{area}}$$

$$Q_{\max} = \sigma_{\max} \cdot 4\pi R^2$$

$$Q_{\max} = \epsilon_{\max} \cdot \epsilon_0 \cdot 4\pi R^2$$

the bigger the radius, the more charge can be stored before breakdown

$$\text{Since } E = \frac{Q}{4\pi\epsilon_0 R^2} + V = \frac{Q}{4\pi\epsilon_0 R} \quad V = E \cdot R$$

$$\text{Then } V_{\max} = E_{\max} \cdot R$$

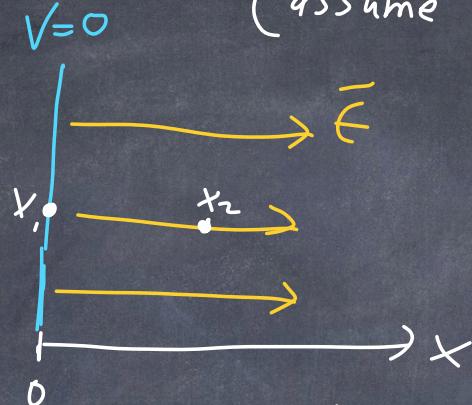
\Rightarrow the maximum potential before discharge increases with $R \Rightarrow$ big R means higher risk of lightning





what is $V(x)$ if $\bar{E} = 10 \frac{N}{C} \hat{x}$?
 (assume that $V=0$ at $x=0$)

$$\bar{E} = 10 \frac{N}{C} = 10 \frac{V}{m}$$



$$dV = -\bar{E} \cdot d\ell$$

$$dV = -10 \frac{V}{m} \hat{x} \cdot dx \hat{x}$$

$$dV = -10 dx \left[\frac{V}{m} \right]$$

$$V(x_2) - V(x_1) = \int_{x_1}^{x_2} dV = \int_{x_1}^{x_2} -10 dx \left[\frac{V}{m} \right] = -10 \left[x \right] \left[\frac{V}{m} \right]_{x_1}^{x_2}$$

$$= 10 \frac{V}{m} (x_1 - x_2)$$

we are told that $V=0$ at $x=0$

so $V(x_2) - V(x_1) = 10 \frac{V}{m} (x_1 - x_2)$

since $V(x_1=0) = 0$

$$V(x_2) - 0 = 10 \frac{V}{m} (0 - x_2) \Rightarrow V(x_2) = -10 x_2 \left[\frac{V}{m} \right]$$

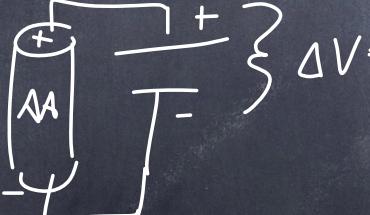
we saw that $\Delta U = q \Delta V$

\uparrow \uparrow \uparrow
 energy charge potential
 [J] [C] [V]

A convenient unit of energy is the electron volt
 example $[e \cdot V]$

$$\Delta U = 1 \text{ eV} = 1 (1.6 \times 10^{-19} \text{ C})(1 \text{ V}) = 1.6 \times 10^{-19} \text{ J}$$

\uparrow
 charge
 of electron

battery  $\Delta V = 1.5 \text{ V} \Rightarrow$ An electron moving through 1.5 V of potential difference will gain 1.5 eV of energy

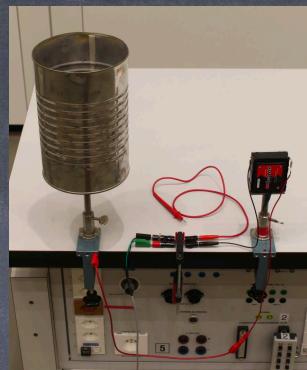
The Large Hadron Collider (LHC) at CERN has protons with energy $6.5 \text{ TeV} = 6.5 \times 10^{12} \text{ eV}$



ES43



ES62



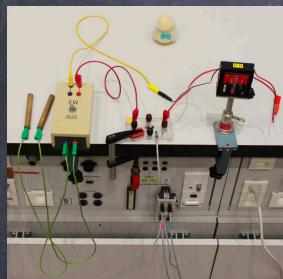
ES12



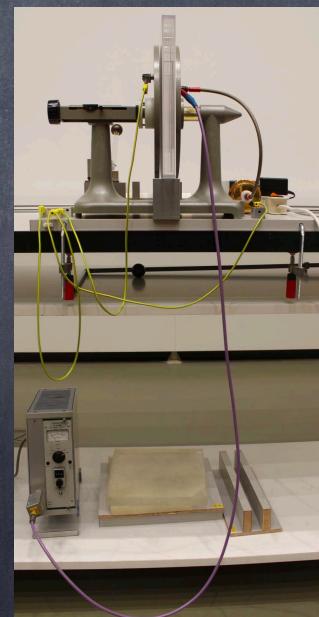
ES28



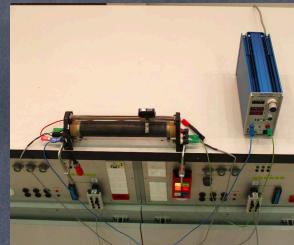
ES20



ES70



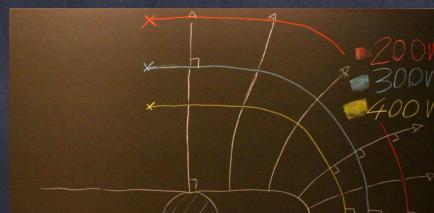
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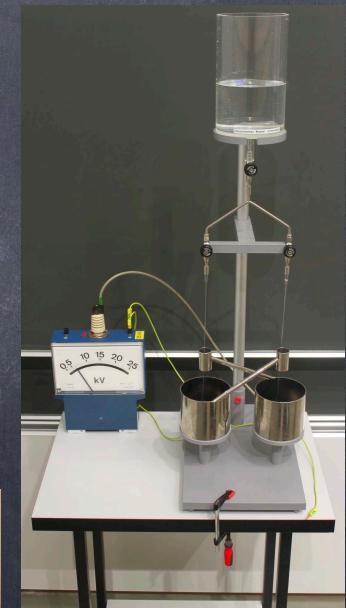
ES61



ES14



ES10



ES25