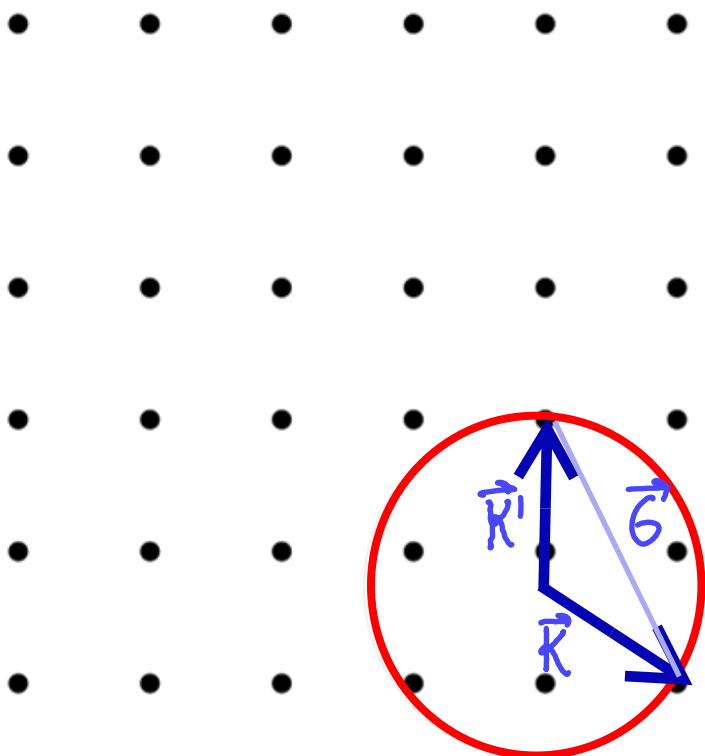


Methods of Scattering experiments

Ewald construction for diffraction

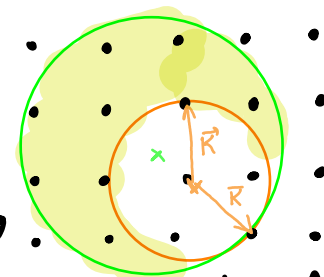


R.L.

$$K = \frac{2\pi}{\lambda}$$

Diffraction condition $\Delta \vec{R} = \vec{G}$

⇒ Probability to obtain a diffraction spot is low

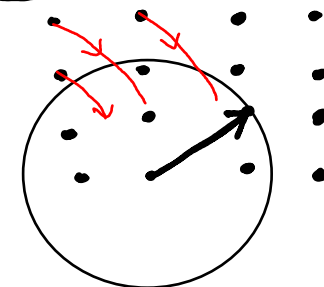


Methods

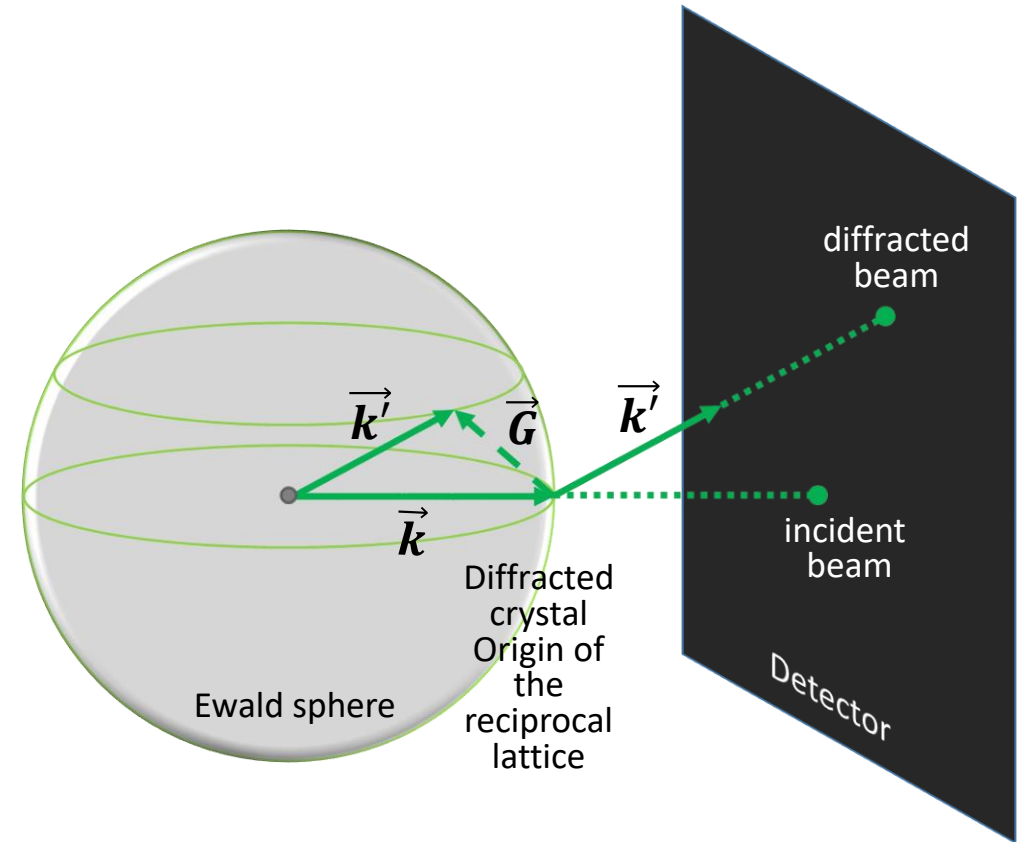
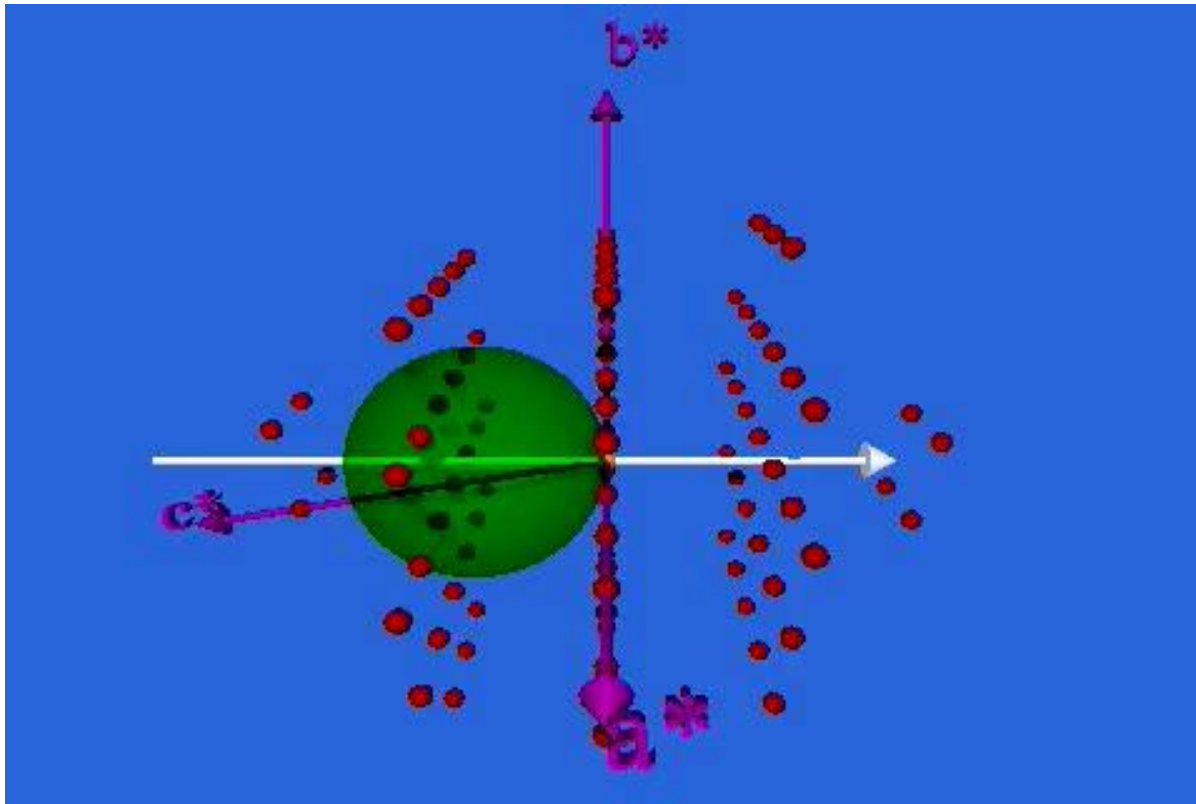
* Laue Method

(range of λ instead of a monochromatic incident wave)

* Rotating Crystal Method



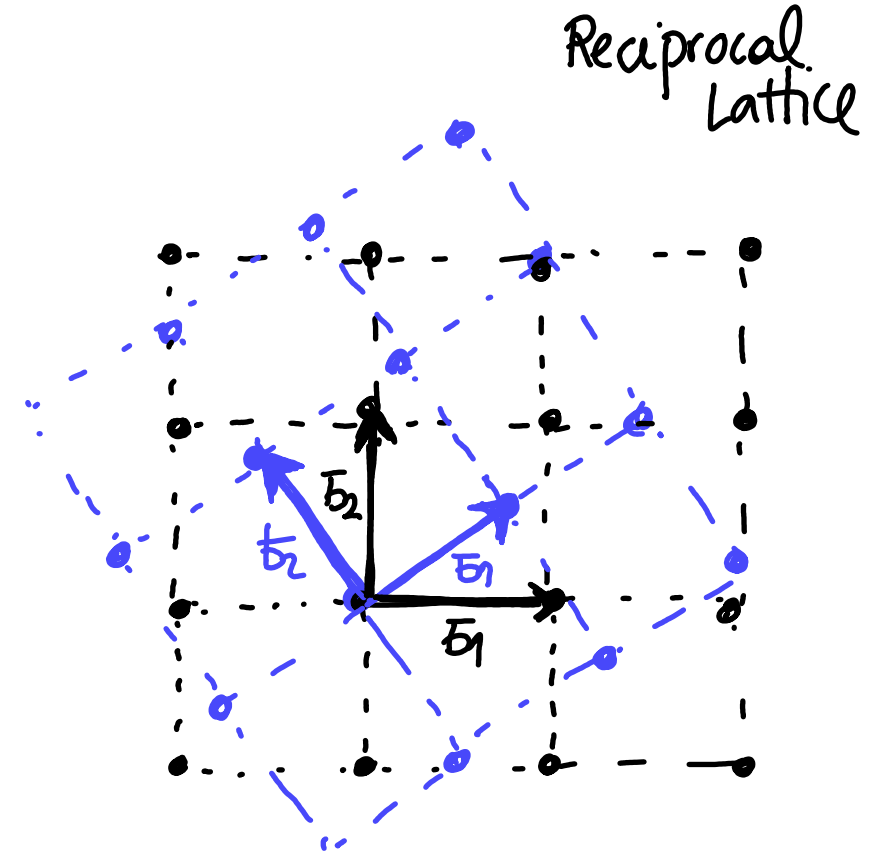
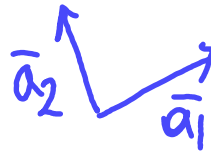
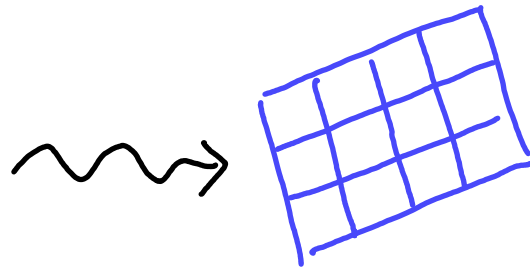
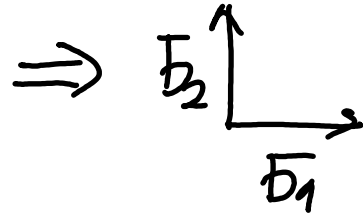
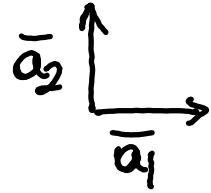
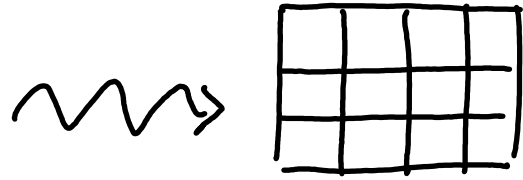
Ewald sphere



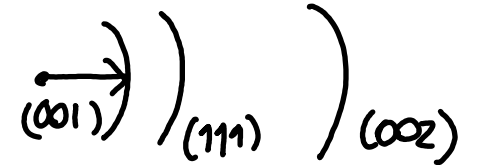
(not single crystalline samples)

Powder diffraction

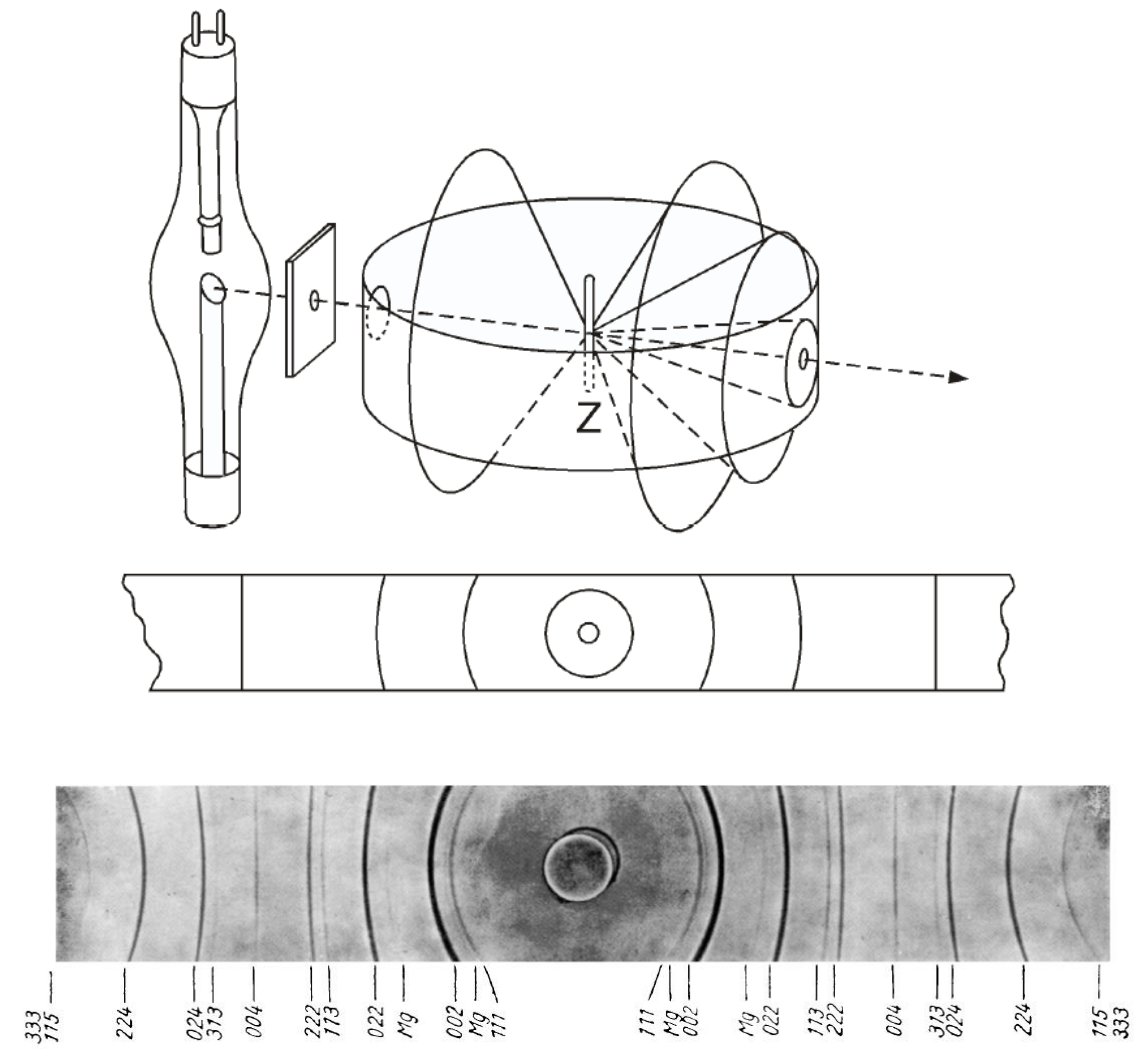
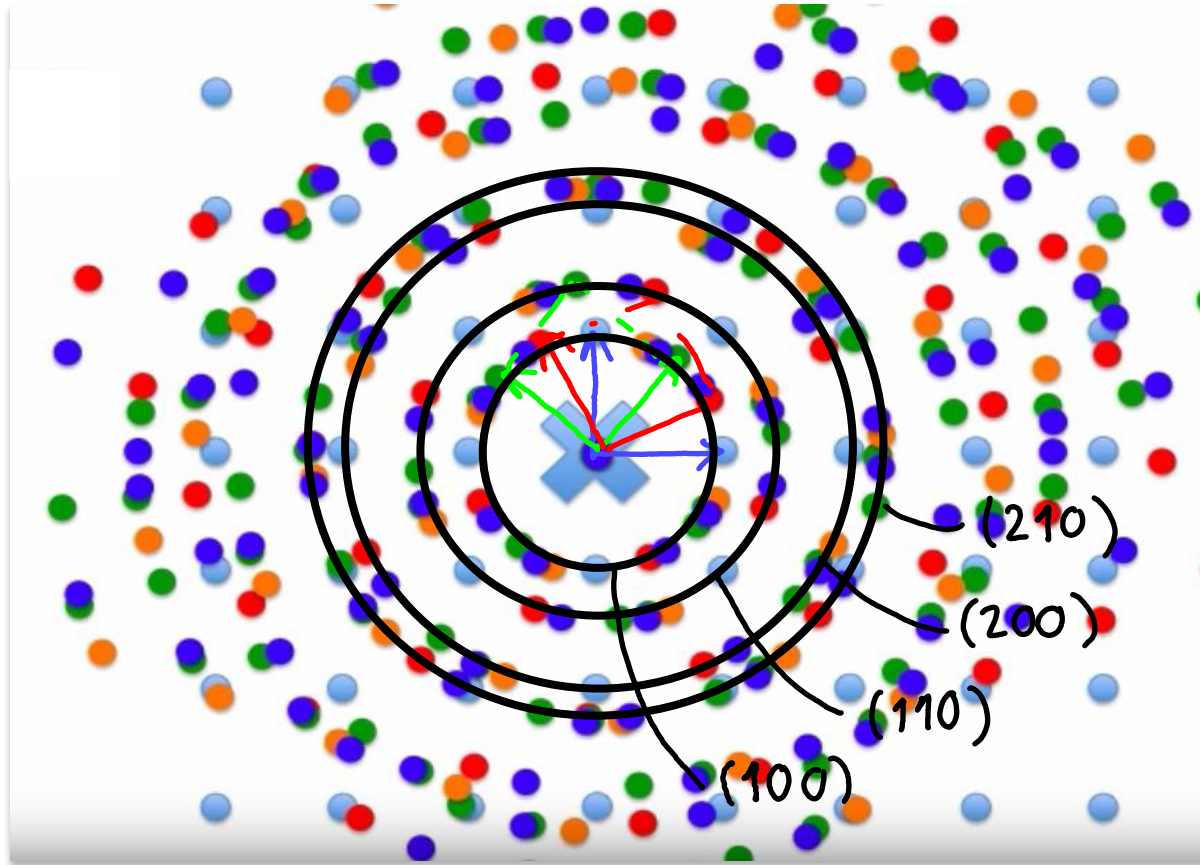
or Debye-Scherrer method



↳ superposition of scattering of many small crystals (which can be oriented in any direction)



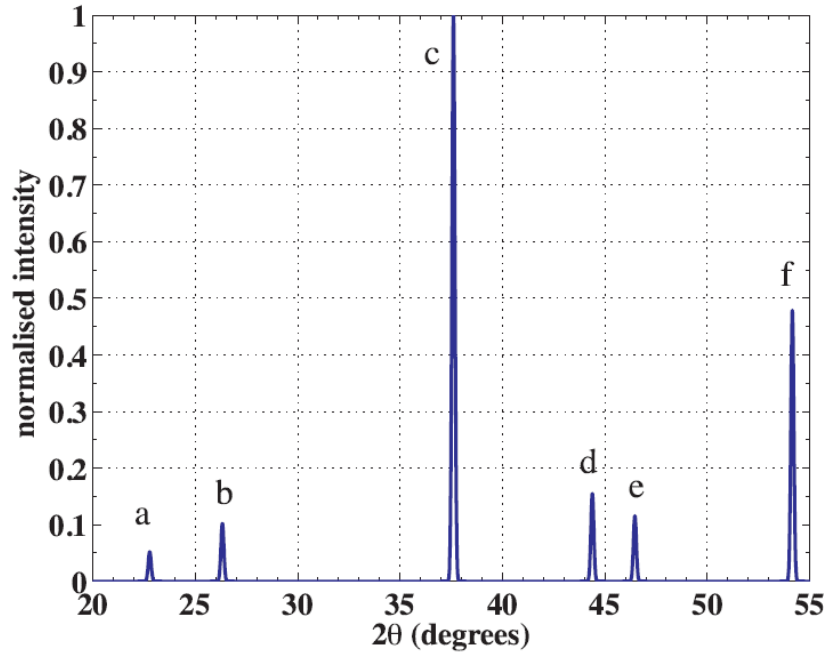
Powder diffraction



Debye-Scherrer-Diagramm von MgO aus Gerthsen, Kneser, Vogel: *Physik*, 13. Aufl. (Springer, Berlin, Heidelberg, New York 1978) Abb. 12.37

Powder diffraction analysis

(some tips for the exercises...)



From Experiment \Downarrow

$$2d \sin \theta = \lambda \Rightarrow d = \frac{\lambda}{2 \sin \theta}$$

We Know

* Lattice Plane selection rules : table

$\{hkl\}$	cubic	bcc	fcc	N
100				↖
110				
⋮				

* Assuming a cubic lattice :

$$d(hkl) = \frac{\lambda}{2 \sin \theta} = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

$$\frac{a^2}{d^2} = h^2 + k^2 + l^2 \equiv N$$

* Remember multiplicity :

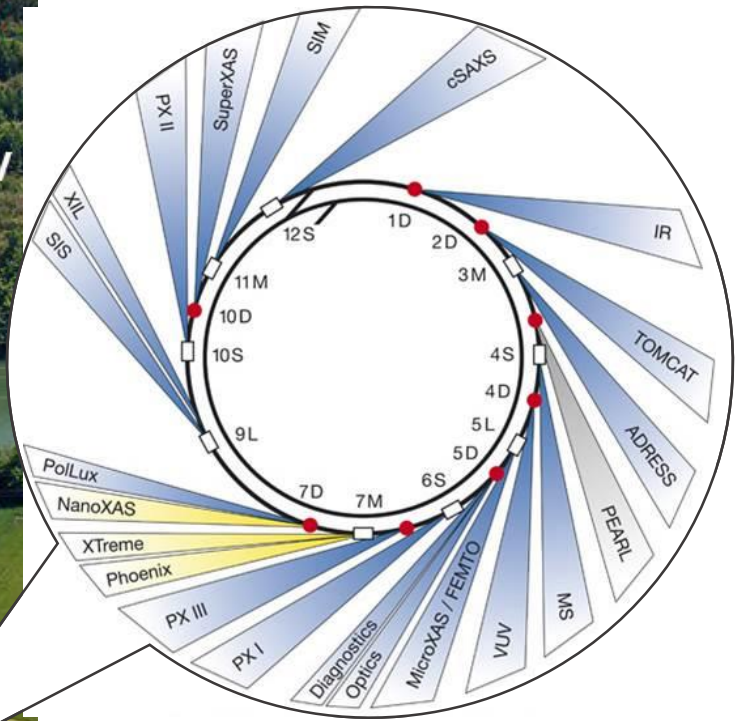
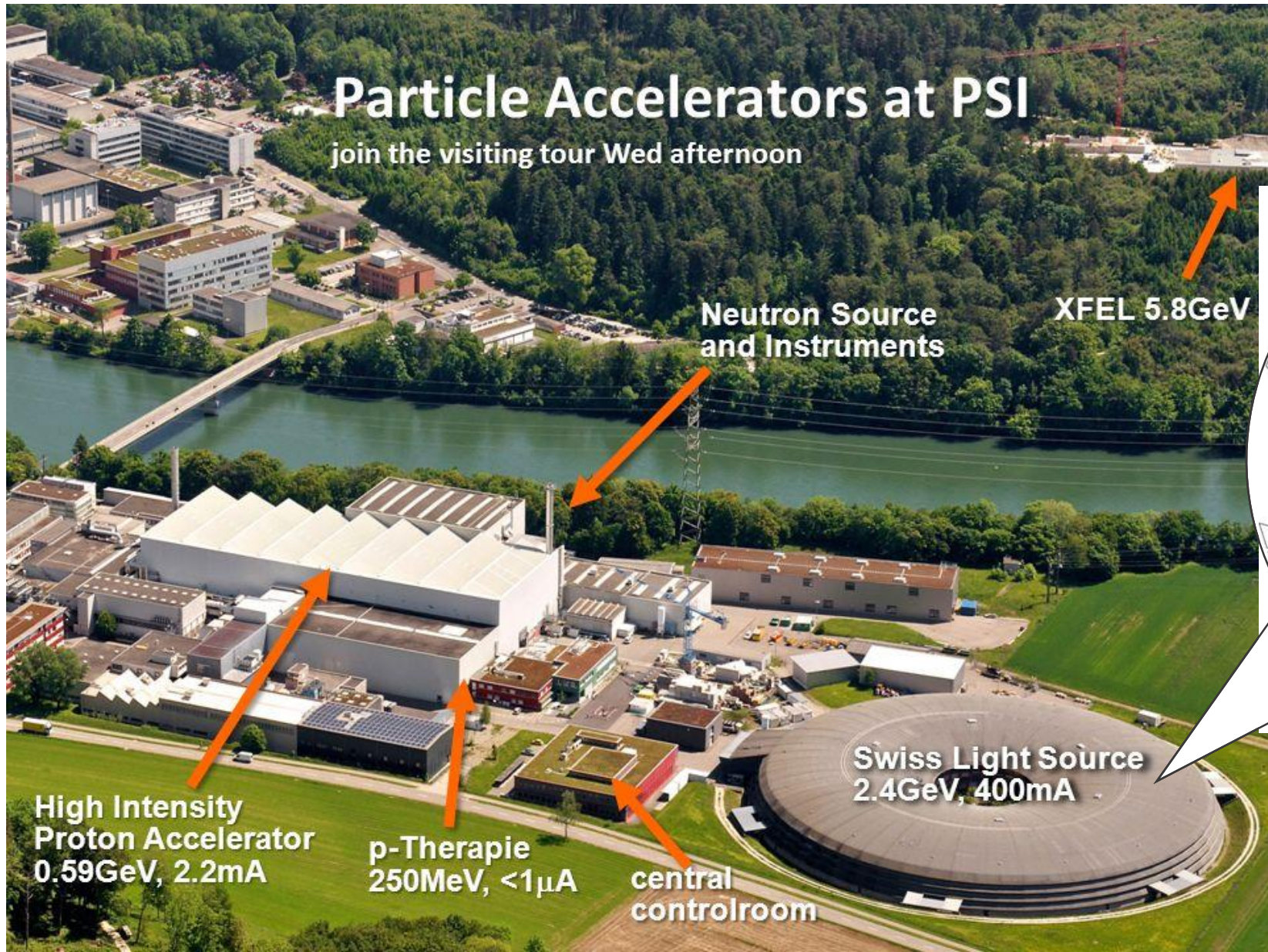
$$I_{\{hkl\}} \propto M_{\{hkl\}} |S_{hkl}|^2$$

X-ray diffractometer



Particle Accelerators at PSI

join the visiting tour Wed afternoon



Learning outcomes - Scattering

- Understand diffraction of waves from crystals in both Laue and Bragg formulations

$$2d \sin\theta = n\lambda \quad \Delta\vec{k} = \vec{G}$$

- Brillouin zone and its special role in diffraction
- Structure factor, atomic form factor
- There are systematic absences in diffraction peaks depending on the crystal structure
- Analyze a powder diffraction pattern (see exercises!)